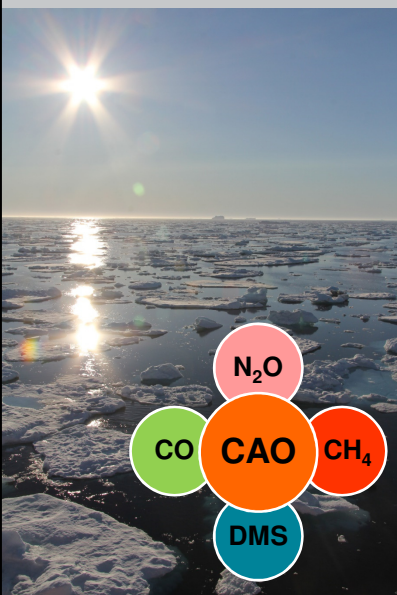


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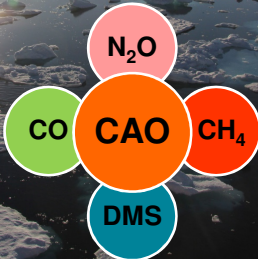

PETRA



**Pathways and emissions of climate-relevant trace gases in a changing Arctic Ocean – PETRA**

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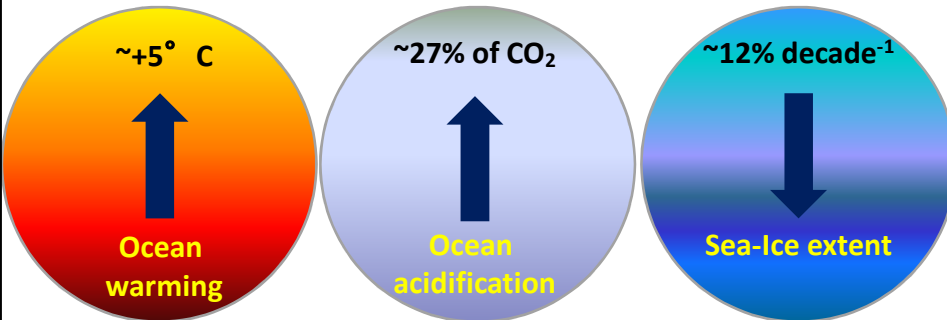
Ian Brown, Hanna Campen, Moritz Baumann, Glen Tarran, Jackie Maud, Damian Arévalo-Martínez, Gennadi Lessin, Vas Kitidis, Yuri Artioli, Darren Clark, Tina Baustian

GEOMAR Helmholtz Centre for Ocean Research Kiel

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**Warming, Acidification, Decreasing Sea-Ice**



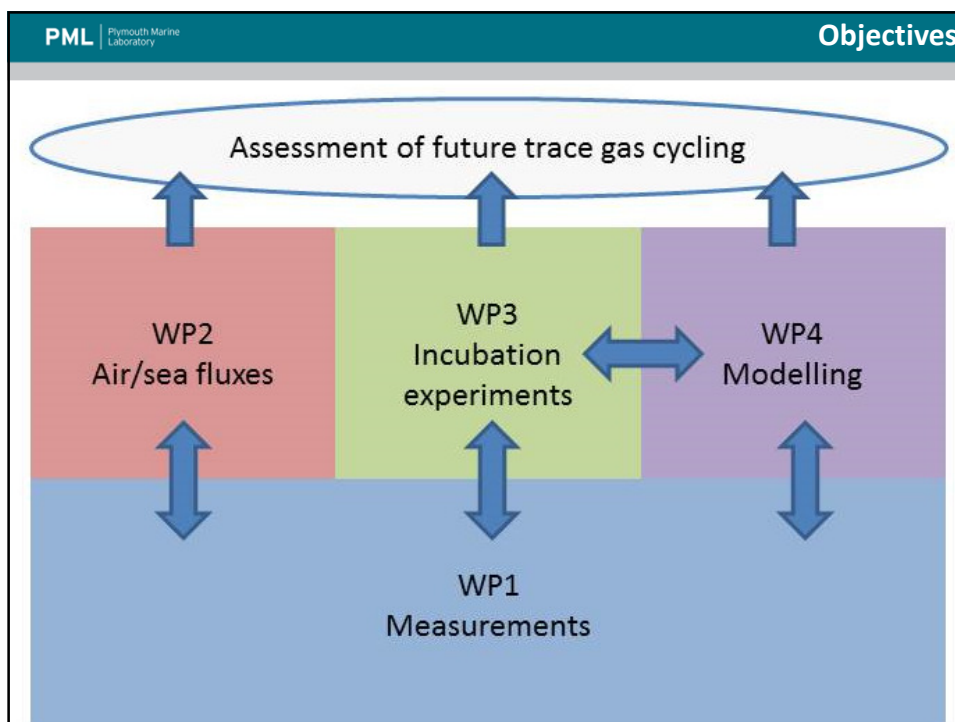
**~+5° C**  
**↑**  
**Ocean warming**

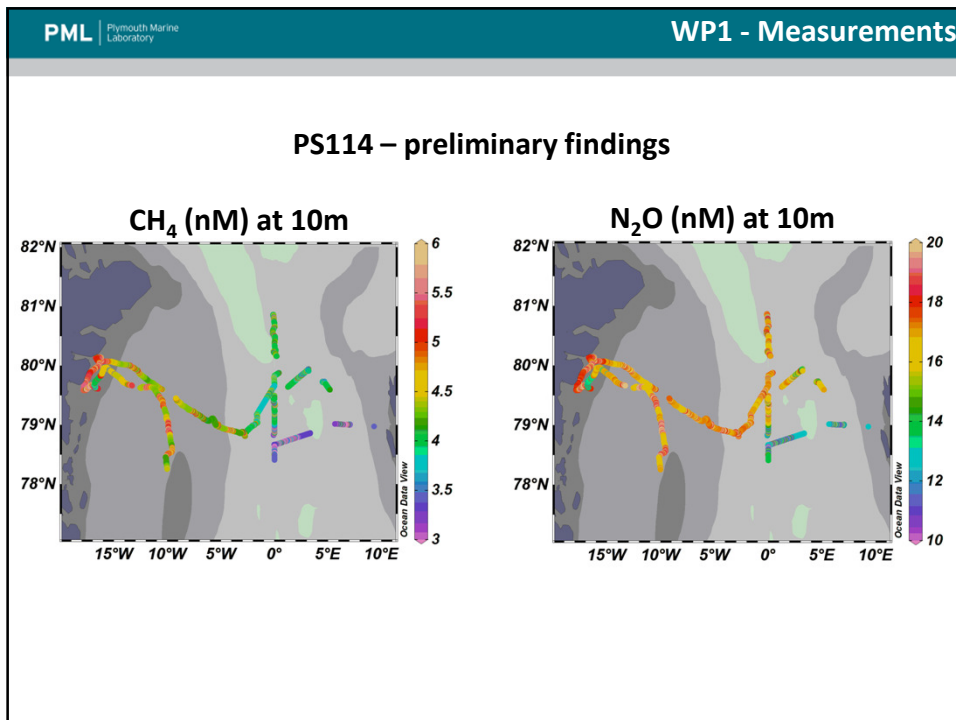
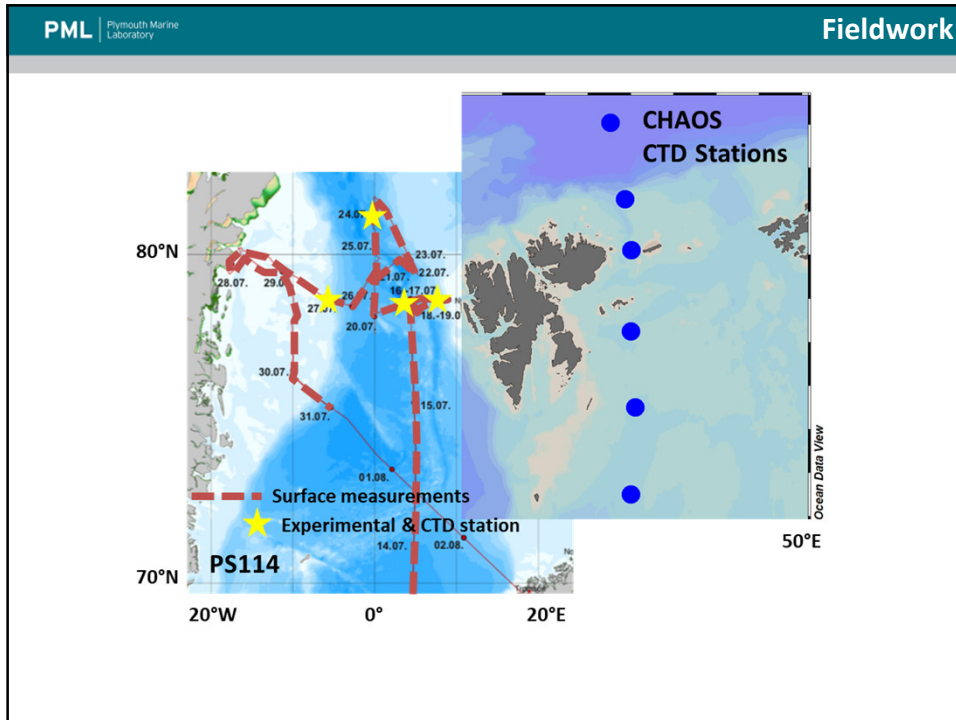
**~27% of CO<sub>2</sub>**  
**↑**  
**Ocean acidification**

**~12% decade<sup>-1</sup>**  
**↓**  
**Sea-Ice extent**

**Combined impact may prove additive, synergistic or antagonistic**

PML Plymouth Marine Laboratory		Rationale		
	T° C	OA	Light	Most vulnerable processes in the Arctic Ocean
N <sub>2</sub> O	X	X		Nitrification
CH <sub>4</sub>	X	X		Aerobic CH <sub>4</sub> oxidation, non-conventional CH <sub>4</sub> production via DMSP
DMS	X	X	X	(Multiple)
CO	X		X	photochemical production from POM and CDOM





PML Plymouth Marine Laboratory WP2 – Air-Sea fluxes

**Computation of the air-sea gas exchange in ice-covered regions:**

$$F = k_{\text{eff}} \Delta C = (1-f) k_{\text{ice}} + f k$$

**A graphic illustration of the mechanisms that can lead to turbulence production and gas exchange in the ice–ocean boundary layer.**  
(Fig. from Loose et al., 2014)

PML Plymouth Marine Laboratory WP3 – Incubation experiments

**PS114 – preliminary findings**

**Time series of [NH<sub>4</sub><sup>+</sup>] at ambient temperature**

Expt-04 Amb

Hours	Control pH = 8.13 (nM)	RCP 6.0 pH = 7.67 (nM)	RCP 8.5 pH = 7.52 (nM)
0	100	100	100
24	150	150	200
48	150	150	200
72	150	200	400
96	150	350	650

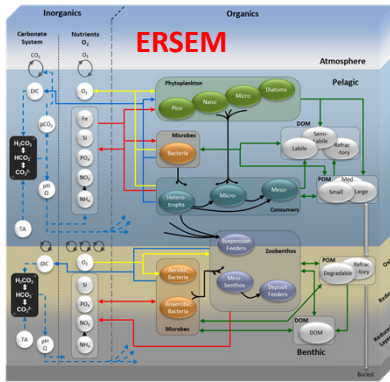
- Control pH = 8.13
- RCP 6.0 pH = 7.67
- RCP 8.5 pH = 7.52

**Summary of N<sub>2</sub>O incubations at 48 hours**

Condition	Expt-01-Amb	Expt-01 +2°C	Expt-02-Amb	Expt-02 +2°C	Expt-03-Amb	Expt-03 +2°C	Expt-04-Amb	Expt-04 +2°C
Control	100	100	100	100	100	100	100	100
pH -1 (RCP6.0)	96	97	97	98	98	99	99	100
pH -2 (RCP8.5)	93	94	94	95	95	96	96	97

PML Plymouth Marine Laboratory WP4 - Modelling

### Mechanistic understanding from models



**ERSEM**

**NH<sub>4</sub>**

$NH_4 \xrightarrow{O_2 \text{ limitation}} NO_3^-$   
 $\downarrow$   
 $0.5N_2O$

Lessin et al., in prep

**DMS**

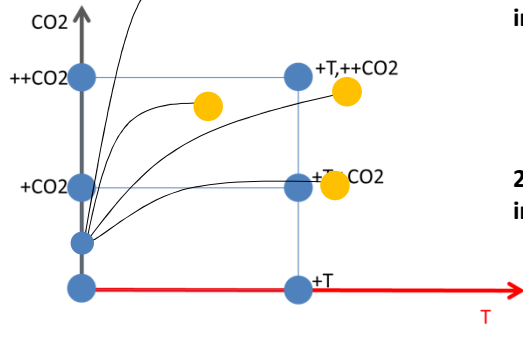
DMSP  $\xrightarrow{P_{exu}}$  DMS  $\xrightarrow{Photo}$  DOC  
 DMSP  $\xrightarrow{B_{exu}}$  B  $\xrightarrow{B_{prod}}$  DMS  $\xrightarrow{B_{upt}}$  B

Polimene et al., Biogeochemistry, 2011

**Models for CO and CH<sub>4</sub> are being developed**

PML Plymouth Marine Laboratory WP4 - Modelling

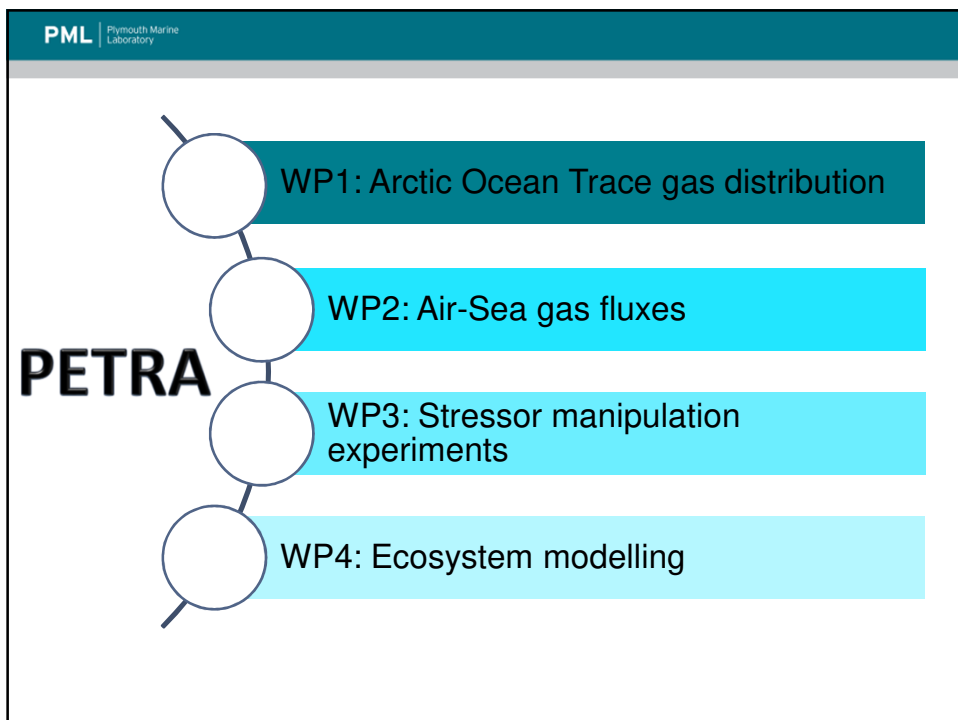
### Mechanistic understanding from models



1. experimental data to define single and multiple stressor functional response and assess stressors interaction
2. model to explore the potential impact in the wider stressor space

PML Plymouth Marine Laboratory PETRA will deliver

- High resolution spatial distribution of trace gases**
  - N<sub>2</sub>O, CH<sub>4</sub>, DMS, CO, CO<sub>2</sub>
- Improved air-sea fluxes in regions of sea-ice**
  - Shipboard and satellite measurements
- Experimental evidence of CAO impact on trace gas fluxes**
  - Sensitivity of biogeochemical processes, nutrient chemistry and organism response
- Mechanistic understanding of stressor impacts on trace gases**
  - Including interaction between stressors



$$F = k_{\text{eff}} \Delta C = (1-f) k_{\text{ice}} + f k$$

Parameter input:  $f$ ,  $u$ ,  $V_0$ ,  $Q$  and profiles of  $T$  and  $S$ ;  
Matlab routines to calculate  $k_{\text{eff}}$  in the sea ice zone are available.

The effective gas transfer velocity ( $k_{\text{eff}}$ ) vs. the fraction of open water ( $f$ ).  
(Fig. from Loose et al., 2014)

